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Chemistry Classwork

1. If 10.0 g of K_2CO_3 is dissolved in 100 grams of water, what is the molality of the resulting solution?

2. What will the concentration of the K^+ ions in the above problem?

3. What is the mole fraction of the potassium carbonate in the first problem?

4. What would the vapor pressure of the solution made in #1 if the solution was at 80°C?

5. Below draw a heating curve for a pure substance. Then draw a heating curve for the same substance but with a solute dissolved into the substance.

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